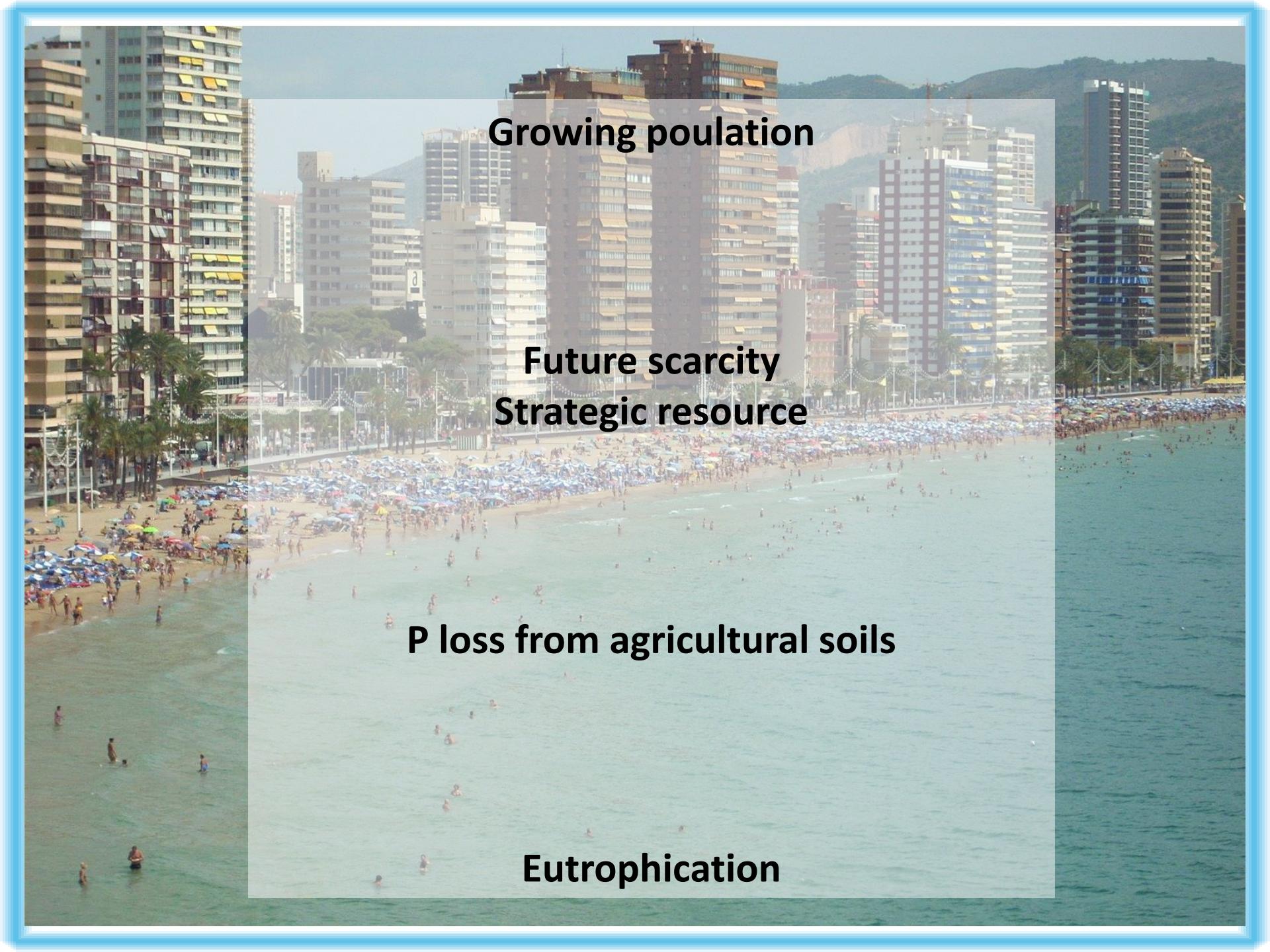


Problemáticas de Fertilización Fosfórica excesiva en Cítricos





Growing poulation

**Future scarcity
Strategic resource**

P loss from agricultural soils

Eutrophication

168Mt phosphate rock (2000)

80% as fertilizer and a small fraction of applied P fertilizer is known to remain available for crops (Delgado et al., 2002; Saavedera et al., 2007)

Only 15% from applied P to final consumer (Cordell et al., 2009).

Demand of P is expected to grow from 50 to 100 % by 2050 (EFMA, 2000; Cordell et al., 2009)

Introduction

**Available P is not a constant fraction of total P;
potential available P for crops is a small fraction of
total P**

between 10 and 70 % of total P in soils in Europe
(Delgado and Torrent, 1997)

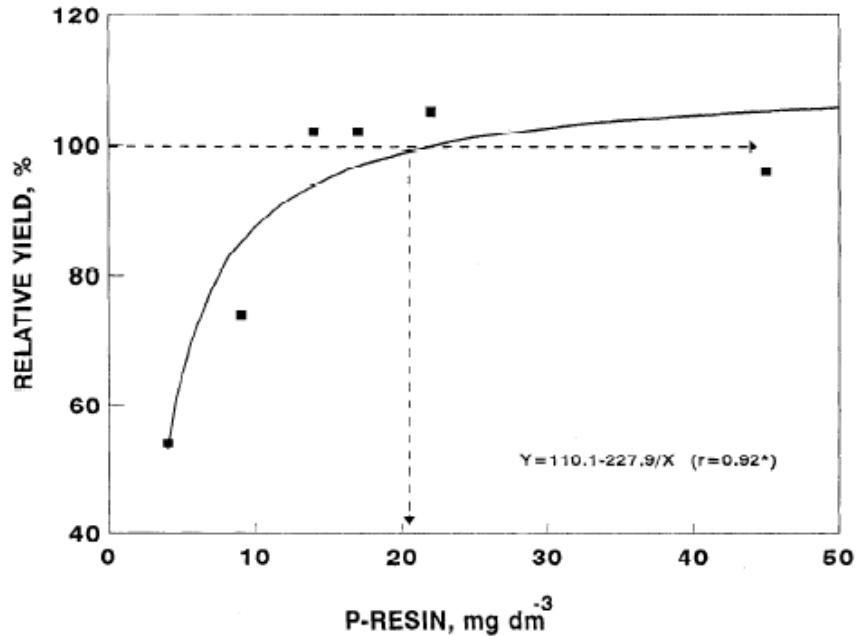
Introduction

Efficient use of phosphorus (P) by plants in agriculture relies on accurate estimation of the phytoavailable soil P.

However, poor relationships are frequently observed between P availability indices, such as Olsen P, and P uptake by plants

Introduction

Phosphorus availability to plants is mainly ruled by the concentration of P in the soil solution ('intensity' factor, I), the amount of P in the solid phase that can be easily made available to plants ('quantity' factor, Q), and the capacity of soil to keep P concentration in soil solution (P buffer capacity, PBC) (Sánchez-Alcalá et al., 2014).

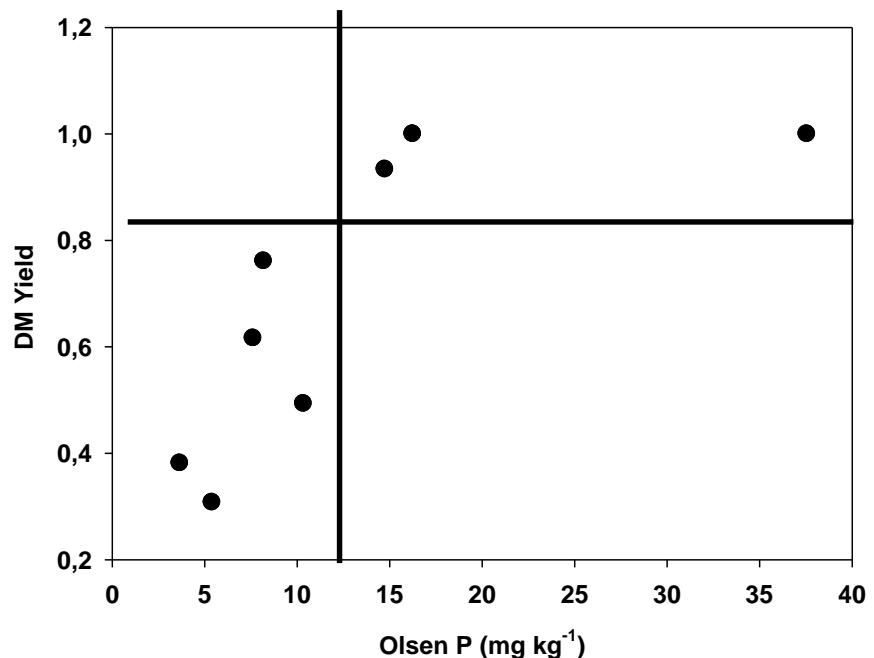


(Recena et al., 2017)

Phosphorus and potassium soil test and nitrogen leaf analysis as a base for citrus fertilization

J. A. Quaggio, H. Cantarella and B. van Raij

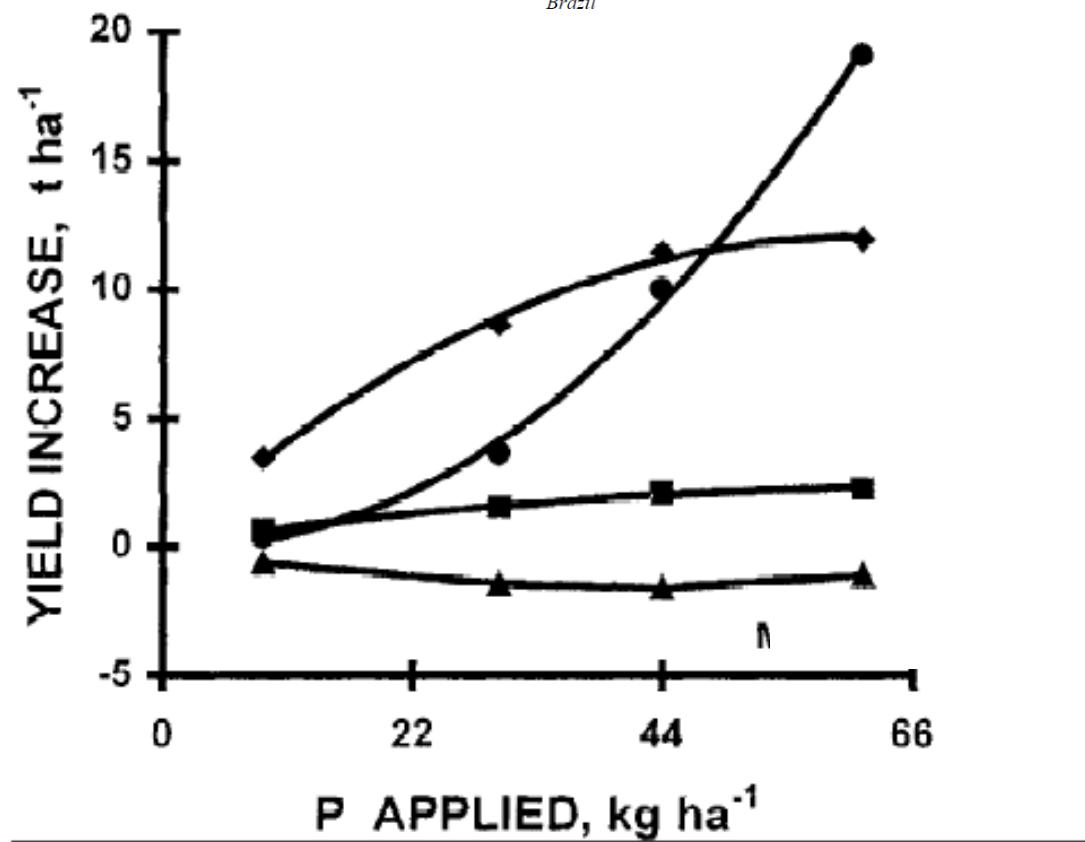
Section of Soil Fertility and Plant Nutrition, Instituto Agronômico, Caixa Postal 28, 13001-970 Campinas, SP, Brazil



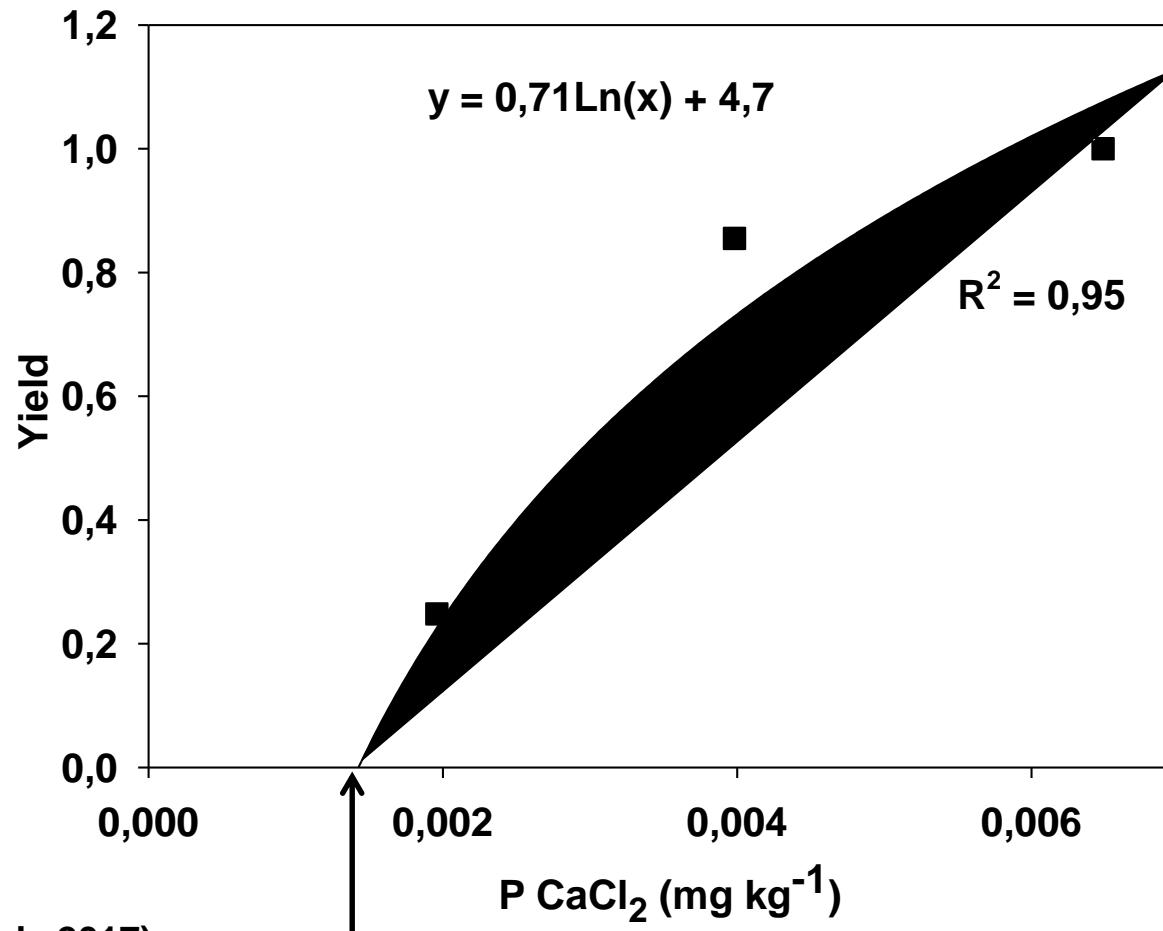
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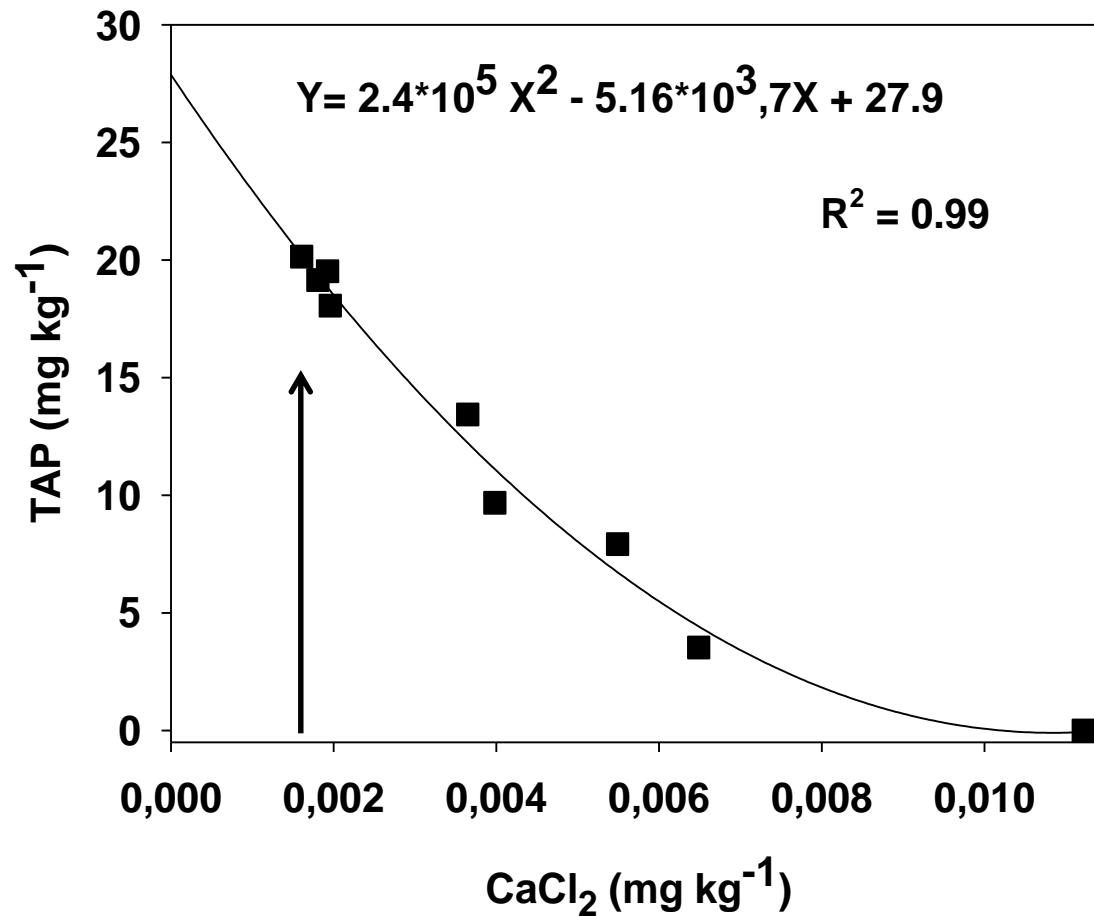
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Threshold values for P CaCl₂ at null P uptake

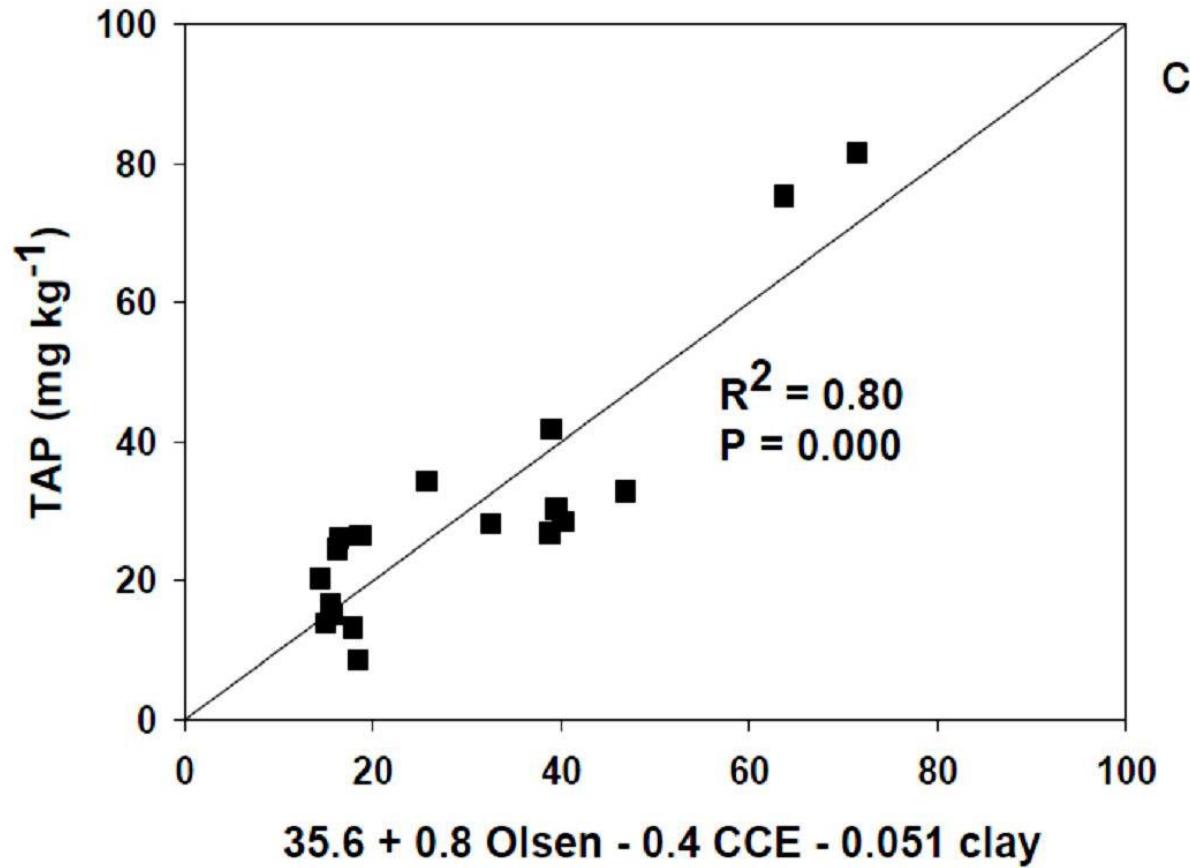


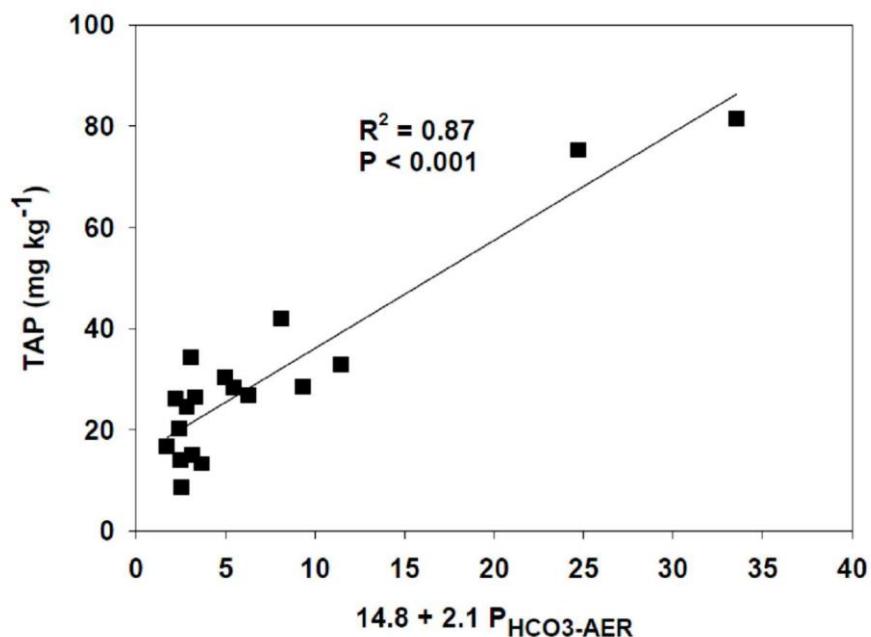
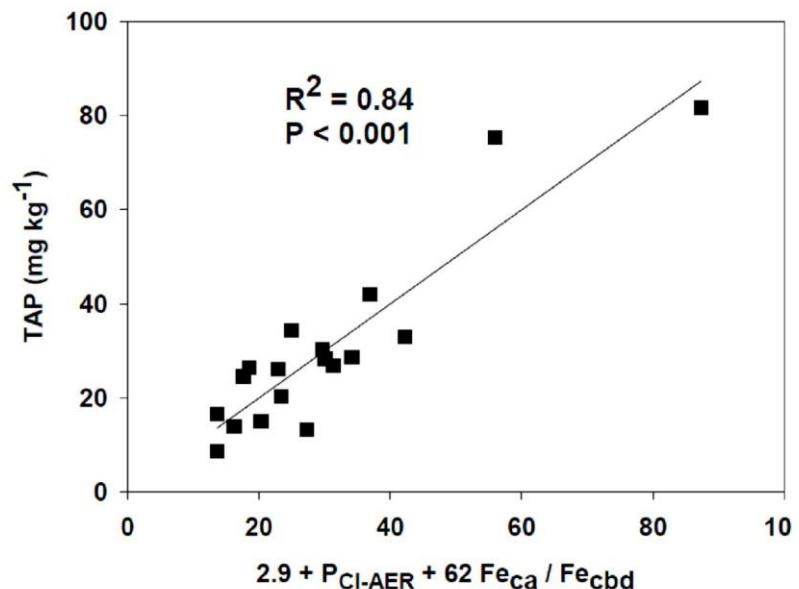
Total available phosphorus (TAP)



(Recena et al., 2017)

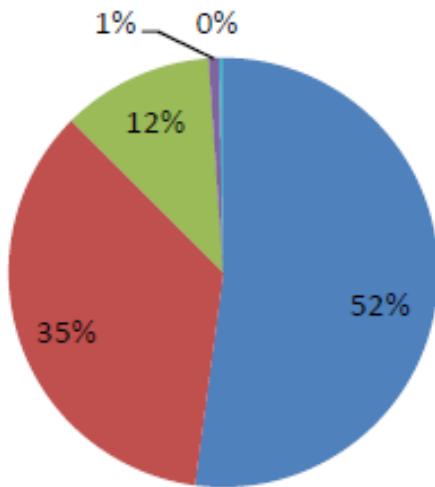
Total available phosphorus (TAP)





Producción de cítricos en España

■ naranja dulce ■ peq.cítricos ■ limón ■ pomelo ■ otros cítricos



Fte: Anuario estadística MAGRAMA

Ud. X 1.000 ha. X 1.000 t

MEDIAS	
superficie	producción
348	13.079
314	5.781
140	1.348
59	1.092
22	192
536	8.578
921	24.628
637	450
9	369
80	2.317

Hortalizas (no patata)
F. cítricos
F. hueso
F. pepita
F tropical
Total frutas (no plátano, no cáscara)
Total frutas y hortalizas
F cáscara
Plátano
Patata

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G. R. Dixon, D. E. Aldous (eds.), *Horticulture: Plants for People and Places, Volume 1*,
DOI 10.1007/978-94-017-8578-5_6, © Springer Science+Business Media Dordrecht 2014

Tree age (years)	Phosphorus (P_2O_5)	
	g/tree	Kg/ha
1–2	0–20	0–8
3–4	30–40	12–16
5–6	50–60	20–24
7–8	80–100	32–40
9–10	120–150	48–60
>10	150–200	60–80

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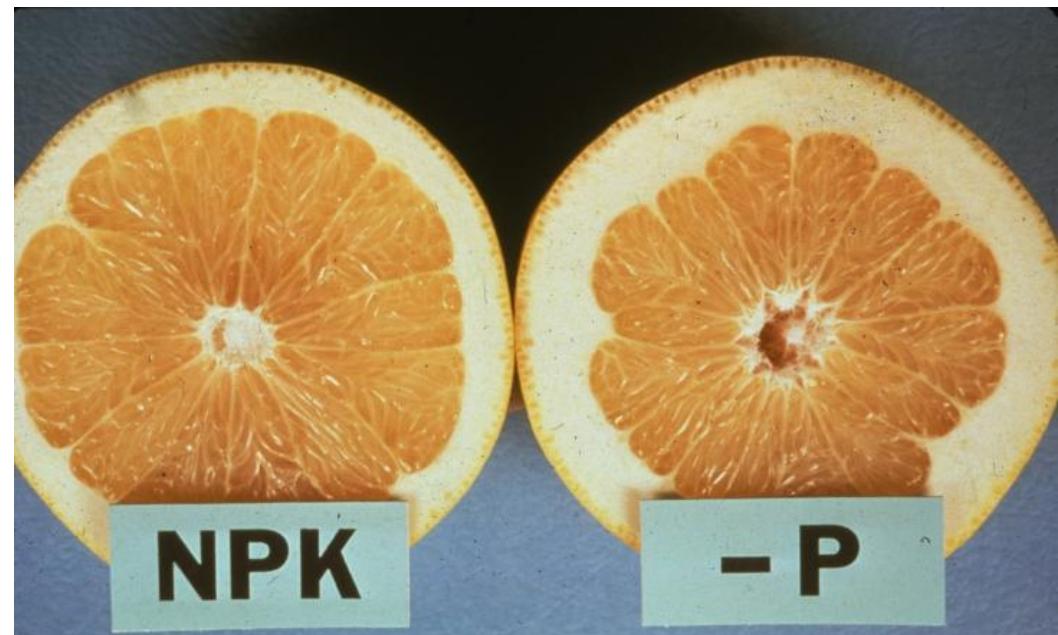
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Leaves analysis guide for diagnosing macronutrient status of citrus adult trees. (Legaz et al. 1995)

Ranges (dry matter basis; %)

		Deficient	Low	Optimum	High	Excess
Sweet orange	N	<230	2.30–2.50	2.51–2.80	2.81–3.00	>3.00
	P	<0.10	0.10–0.12	0.13–0.16	0.17–0.20	>0.20
	K	<0.50	0.50–0.70	0.71–1.00	1.01–1.30	>1.30
Clementine Mandarin	N	<2.20	2.21–2.40	2.41–2.70	2.71–2.90	>2.90
	P	<0.09	0.09–0.11	0.12–0.15	0.16–0.19	>0.19
	K	<0.50	0.50–0.70	0.71–1.00	1.01–1.30	>1.30
Satsuma Mandarin	N	<2.40	2.40–2.60	2.61–2.90	2.91–3.10	>3.10
	P	<0.10	0.10–0.12	0.13–0.16	0.17–0.20	>0.20
	K	<0.40	0.40–0.60	0.61–0.90	0.91–1.15	>1.15



1 - Single models for improving estimations of total available phosphorus (TAP) with Olsen P index and P CaCl_2 extraction

2 - Properties contribuiting to explain the equilibrium between P soild and liquid soild phase are very relevant explaining the accuracy of SPT

3 - TAP estimation with Olsen-P, is improved by PBC and $\text{Fe}_{\text{ca}}/\text{Fe}_{\text{cbd}}$

4 - P sinks provide better estimation of TAP than chemical extractions

5 - AER in Cl^- form gave worse explications of the TAP variance than HCO_3^- ; Cl^- inhibit the exchange reaction (Mayers et al., 2005) and HCO_3^- seems to have similar effect than roots (Sibbensen, 1978).

6 - Overall, P sinks are better correlated with soil factors affecting P availability to plants

-P; macronutriente de concentración variable en la solución del suelo

-con gran afinidad por las superficies de los óxidos de Fe (Torrent, 1987)

-facilidad para formar complejos insolubles, reduciendo la solubilidad del Fe y dificultando su absorción por parte de las plantas (Zheng et al., 2009)

Gracias por su atención.

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